

# Zn-Zr (Zinc-Zirconium)

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The Zn-Zr phase diagram in [Massalski2] was assessed by [1992Dut].

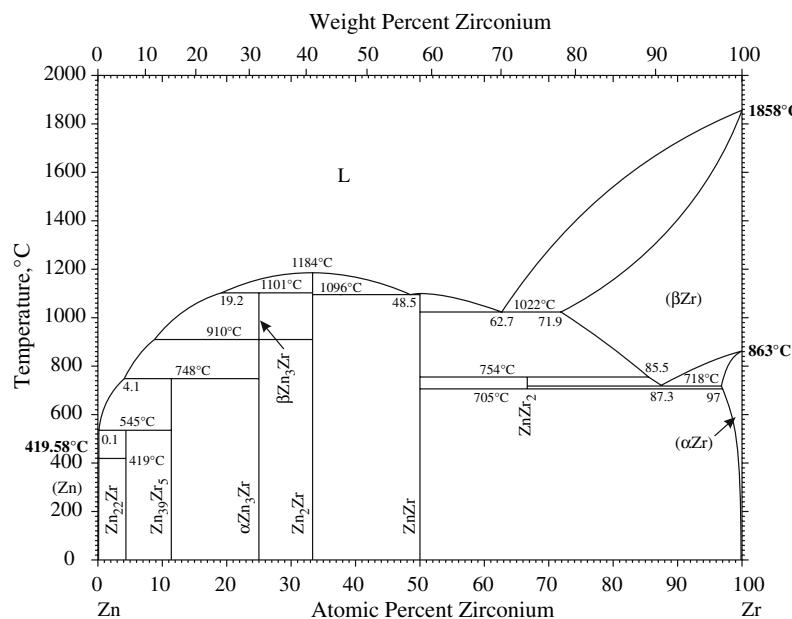
After this phase diagram was published, the existence of the  $\text{ZnZr}_2$  phase was discovered by [2004Wil]. This phase exists in a narrow temperature range from  $750 \pm 24$  and  $706 \pm 6$  °C according to optical metallography, SEM, and TEM investigations. This system was thermodynamically assessed by [2006Arr1]. Three possible thermodynamic models were considered. Because all calculated results are similar, the phase diagrams corresponding to their first model are shown in Fig. 1 and 2. Figure 1 shows the Zn-Zr phase diagram under constrained vapor conditions. The stability of various Zn-Zr phases was discussed by [2006Arr2].  $\text{Zn}_{14}\text{Zr}$  and  $\text{Zn}_6\text{Zr}$  shown in the diagram of [1992Dut] are  $\text{Zn}_{22}\text{Zr}$  and  $\text{Zn}_{39}\text{Zr}_5$ , respectively, according to [2006Arr2]. In addition to the phases in Fig. 1,  $\text{Zn}_2\text{Zr}_3$  may exist at high temperatures ( $>1000$  °C). [2006Arr1]

shows a phase diagram including  $\text{Zn}_2\text{Zr}_3$ . Figure 2 shows the Zn-Zr phase diagram at 1 bar.

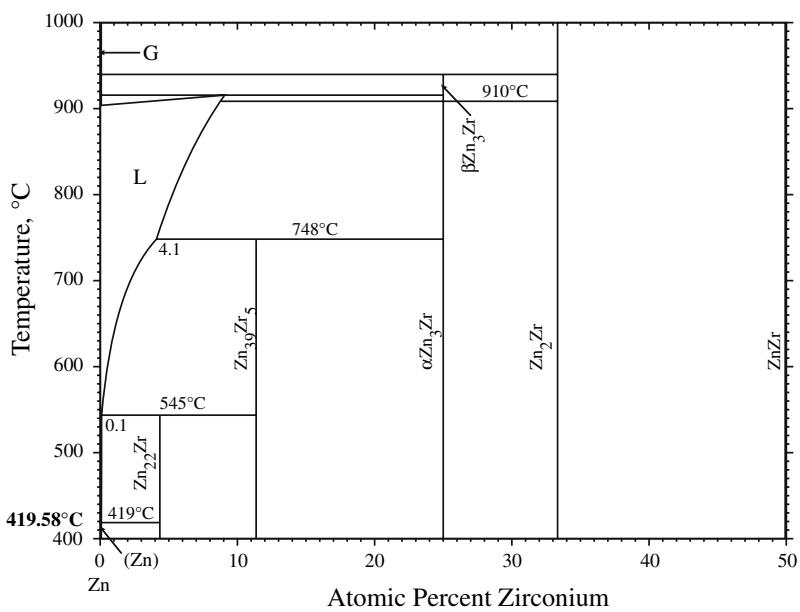
Zn-Zr crystal structure data are given in Table 1.

## References

- 1992Dut:** J. Dutkiewicz, The Zn-Zr (Zinc-Zirconium) System, *J. Phase Equilib.*, 1992, **13**(4), p 430-433
- 2004Wil:** M.E. Williams, W.J. Boettinger, and U.R. Katner, Contribution to the Zr-Rich Part of the Zn-Zr Phase Diagram, *J. Phase Equilib. Diffusion*, 2004, **25**(4), p 355-363
- 2006Arr1:** R. Arroyave and Z.K. Liu, Thermodynamic Modeling of the Zn-Zr System, *Calphad*, 2006, **30**(1), p 1-13
- 2006Arr2:** R. Arroyave, A. van de Walle, and Z.K. Liu, First-Principles Calculations of the Zn-Zr System, *Acta Mater.*, 2006, **54**, p 473-482



**Fig. 1** Zn-Zr phase diagram under constrained vapor conditions

**Fig. 2** Zn-Zr phase diagram at 1 bar**Table 1** Zn-Zr crystal structure data

Phase	Composition at.% Zr	Pearson symbol	Space group	Struktur-bericht designation	Prototype
(Zn)	0	<i>hP2</i>	<i>P6<sub>3</sub>/mmc</i>	<i>A3</i>	Mg
$Zn_{22}Zr$	4.3	<i>cF184</i>	<i>Fd<sub>3</sub>m</i>	...	...
$Zn_{39}Zr_5$	11.4	<i>mC88</i>	<i>C2/m</i>	...	...
$\beta Zn_3Zr$	25	...	...	...	...
$\alpha Zn_3Zr$	25	<i>t**</i>	...	...	...
$Zn_2Zr$	33.3	<i>cF24</i>	<i>Fd<sub>3</sub>m</i>	<i>C15</i>	$Cu_2Mg$
$ZnZr$	50	<i>cP2</i>	<i>Pm <math>\overline{3}m</math></i>	<i>B2</i>	$CsCl$
$ZnZr_2$	66.7	<i>tI6</i>	<i>I4/mmm</i>	<i>D0<sub>22</sub></i>	$MoSi_2$
( $\beta Zr$ )	71.9-100	<i>cI2</i>	<i>I<math>\bar{m}</math> <math>\overline{3}m</math></i>	<i>A2</i>	W
( $\alpha Zr$ )	96.8-100	<i>hP2</i>	<i>P6<sub>3</sub>/mmc</i>	<i>A3</i>	Mg